1. Data about the program of study

	2 au acour me program or staal	
1.1	Institution	The Technical University of Cluj-Napoca
1.2	Faculty	Automation and Computer Science
1.3	Department	Mathematics
1.4	Field of study	Computer Science and Information Technology
1.5	Cycle of study	Bachelor of Science
1.6	Program of study/Qualification	Computer Science/ Engineer
1.7	Form of education	Full time
1.8	Subject code	9.

2. Data about the subject

2.1	Subject name				Math	Mathematical analysis II (Integral calculus and differential equations)					
2.2	2 Subject area			Com	Computer Science and Information Technology						
2.3	3 Course responsible/lecturer				Prof.	Prof. dr. Dumitru Mircea IVAN					
2.4	Teachers in cha	arge o	f app	olications		Lect.	ect. Mircea RUS, Lect. Adela CAPATA				
2.5	Year of study	Ι	2.6	Semester	2	2.7 Assessment exam 2.8 Subject category DF/OF			DF/OB		

3. Estimated total time

Sem.	Subject name	Lecture	ure Applications		Lecture Applications			Individual study	TOTAL	Credit		
		[hours / week.]		[hours / semester]								
			S	L	Р		S	L	Р			
2	Mathematical analysis II (Integral calculus and differential equations)	2	2	-	•	28	28	I	I	98	154	6

3.1	Number of hours per week	4	3.2	of which, course	2	3.3	applications	2
3.4	Total hours in the teaching plan	56	3.5	of which, course	28	3.6	applications	28
Individual study Ho								
Man	ual, lecture material and notes, bibliograp	hy						40
Supp	lementary study in the library, online and	in the fie	ld					14
Prepa	aration for seminars/laboratory works, how	mework, 1	reports	, portfolios, essays				41
Tuto	ring							0
Exam	s and tests							3
Other activities							0	
3.7 Total hours of individual study 98								

3.8	Total hours per semester	154
3.9	Number of credit points	6

4. Pre-requisites (where appropriate)

4.1	Curriculum	Basic knowledge Integral Calculus
4.2	Competence	Competences in elementary Integral Calculus: primitives, definite integrals.

5. Requirements (where appropriate)

-	······································						
5.1	For the course	Videoprojector					
5.2	For the applications	Videoprojector					

	C1 – Operating with basic Mathematical, Engineering and Computer Science concepts						
_ s	C1.1 – Recognizing and describing concepts that are specific to the fields of calculability, complexity,						
nce	programming paradigms, and modeling computational and communication systems						
sic ter	C1.3 – Building models for various components of computing systems						
fes	C1.5 – Providing a theoretical background for the characteristics of the designed systems						
Pro							

Cross npetences	N/A	
com		

7.1	General objective	A presentation of the concepts, notions, methods and fundamental
		techniques used in integral calculus.
7.2	Specific objectives	Use of the integral calculus in order to solve problems in engineering.

8. Contents

8.1. L	ecture (syllabus)	Teaching methods	Notes				
1	Ordinary differential equations (ODE) of order one	Explanation					
2	Linear homogeneous ODE with constant coefficients						
3	Linear non-homogeneous ODE with constant coefficients	Demonstration					
4	Positive and linear functionals.						
5	Riemann-Stieltjes integral. Primitives.	Collaboration					
6	Improper integrals.						
7	Integrals depending on parameters.	Interactive					
8	Special functions	activities					
9	Paths. Vector fields. Line integrals with respect to the coordinates. Circulation.						
10	Differential Forms. Exact differential forms. Path-independence. Work.						
11	Line integrals with respect to the arc length. Total mass, center of mass.						
12	Double integral. Green-Riemann formula.						
13	Surface integral. Flux of vector field across a surface. Stokes' Theorem.						
14	Volume integral. Gauss-Ostrogradsky Theorem. MATHEMATICA capabilities.						
Biblio	graphy		•				
1	. Mircea Ivan. Elemente de calcul integral. Mediamira, Cluj-Napoca, 2003. ISBN	973-9357-40-7.					
2	2. Dumitru Mircea Ivan. Calculus. Editura Mediamira, Cluj-Napoca, 2002. ISBN 9	73-9358-88-8.					
8.2. A	Applications (Seminars)	Teaching methods	Notes				
1	Ordinary differential equations (ODE) of order one (Exercises)						
2	Linear homogeneous ODE with constant coefficients (Exercises)						
3	Linear non-homogeneous ODE with constant coefficients (Exercises)						
4	Positive and linear functionals (Exercises)	Explanation					
5	Riemann-Stieltjes integral. Primitives (Exercises)	1					
6	Improper integrals (Exercises)	Demonstration					
7	Integrals depending on parameters (Exercises)						
8	Special functions (Exercises)	Collaboration					
9	Line integrals with respect to the coordinates (Exercises)						
10	Differential Forms (Exercises)	Interactive					
11	Line integrals with respect to the arc length. (Exercises)	activities					
12	Double integral. Green-Riemann formula. (Exercises)						
13	Surface integral. (Exercises)						
14	Volume integral. MATHEMATICA related capabilities. (Exercises)						
Biblic	Bibliography						
1	Dumitru Mircea Ivan, et al. Analiză matematică - Culegere de probleme pentru	seminarii, examene și					
	concursuri. Editura Mediamira, Cluj-Napoca, 2002. ISBN 973-9357-20-2.						
2	2. Mircea Ivan et al. Culegere de Probleme Pentru Seminarii, Examene și Concurs	uri. UT Press, Cluj-Na	ipoca,				
1	2000		-				

9. Bridging course contents with the expectations of the representatives of the community, professional associations and employers in the field

Collaboration with engineers in order to identify and solve problems raised by the market.

10. Evaluation

Activity type	10.1	Assessment criteria	10.2	Assessment methods	10.3	Weight in the final grade	
Course		Abilities of understanding and using creatively the concepts and proofs		Written examination		30%	
Applications		Abilities of solving problems and		Written examination		70%	
		applying algorithms					
10.4 Minimum standard of performance							
Ability to present coherently a theoretical subject and to solve problems with practical content.							

Course responsible Prof.dr. Dumitru Mircea Ivan Head of department Prof.dr.eng. Rodica Potolea

1. Data about the program of study

1.1	Institution	The Technical University of Cluj-Napoca
1.2	Faculty	Automation and Computer Science
1.3	Department	Computer Science
1.4	Field of study	Computer Science and Information Technology
1.5	Cycle of study	Bachelor of Science
1.6	Program of study/Qualification	Computer Science/ Engineer
1.7	Form of education	Full time
1.8	Subject code	10.

2. Data about the subject

2.1	Subject name				Spec	Special Mathematics in Engineering						
2.2	2.2 Subject area				Com	Computer Science and Information Technology						
2.3	2.3 Course responsible/lecturer					Prof.	Prof.dr. Ioan RASA <u>Ioan.Rasa@math.utcluj.ro</u>					
2.4	2.4 Teachers in charge of applications					Conf	Conf. dr. Daniela Inoan - <u>Daniela.Inoan@math.utcluj.ro</u>					
2.5	Year of study	Ι	2.6	Semester	2	2.7	Assessment	exam	2.8	Subject category	DF/OB	

3. Estimated total time

Sem.	Subject name	Lecture	App	olicati	ions	Lecture	re Applications		Individual study	TOTAL	Credit	
		[hours / week.]		[hours / semester]			ter]					
			S	L	Р		S	L	Р			
2	Special Mathematics II	2	2	-	•	28	28	-	-	100	156	6

3.1	Number of hours per week	4	3.2	of which, course	2	3.3	applications	2
3.4	Total hours in the teaching plan	56	3.5	of which, course	28	3.6	applications	28
Indiv	idual study							Hours
Man	ual, lecture material and notes, bibliograp	hy						20
Supp	lementary study in the library, online and	l in the fiel	ld					21
Prepa	aration for seminars/laboratory works, ho	mework, r	reports	s, portfolios, essays				56
Tuto	ring							
Exam	s and tests							3
Other activities								
3.7	Total hours of individual study		100					
3.8	Total hours per semester		156					

4.	Pre-requisites	(where	appropriate)	

Number of credit points

	in the requisites (mere uppropria	
4.1	Curriculum	Elementary knowledge of complex numbers. Elements of calculus.
4.2	Competence	Competences in using complex numbers (in algebraic and trigonometric form).
		Ability to calculate derivatives and real integrals.

6

	5. Requirements (where appropria	te)
5.1	For the course	Blackboard, videoprojector
5.2	For the applications	Blackboard, videoprojector

6. Specific competences

3.9

1	
	C1 – Operating with basic Mathematical, Engineering and Computer Science concepts
– s	C1.1 – Recognizing and describing concepts that are specific to the fields of calculability, complexity,
na	programming paradigms, and modeling computational and communication systems
sio	C1.3 – Building models for various components of computing systems
fes	C1.5 – Providing a theoretical background for the characteristics of the designed systems
Pro	
- 0	

ces	N/A	
Cross		
comj		

7.1	General objective	A presentation of the concepts, notions, methods and fundamental techniques used in complex functions theory and integral transforms theory.
7.2	Specific objectives	Use of the complex functions theory and integral transforms theory for solving problems in engineering.

8. Contents

8.1. Lecture (syllabus)	Teaching methods	Notes
1 Complex numbers. Operations, topology in C.	Explanation	
2 Continuity. Monogenic functions. The Cauchy-Riemann conditions.		
Holomorphic functions.	Demonstration	
3 The complex integral. Definition. Cauchy's integral theorem. Cauchy's integral		
formula.	Collaboration	
4 Taylor and Laurent series. Singular points, classification.		
5 Residues. The Residue Theorem.	Interactive	
6 Applications of the Residue Theorem.	activities	
7 Real integrals calculated with complex methods.		
8 The Fourier transform. Definition, properties.		
9 Applications of the Fourier transform.		
10 The Laplace transform. Definition and properties.		
11 The inverse Laplace transform.		
12 Applications of the Laplace transform.		
13 The z transform. Applications.		
14 Difference equations. The z transform applied to solving difference equations.		
Bibliography		
1. A.I. Mitrea, Analiza matematica in complex (curs+culegere de probleme), Ed. Mediam	ira, Cluj-Napoca, 200	5.
2. A.I. Mitrea, Transformari integrale si discrete (curs + culegere de probleme) Ed. Media	mira, Cluj-Napoca, 20	004.
3. M.L. Krasnov, A.I. Kiselev, G.I. Makarenko, Functions of a Complex Variable, Operat	ional Calculus and Sta	ability
Theory, Mir Publishers, Moscow, 1984.	I	1
8.2. Applications (Seminars)	Teaching methods	Notes
1 Operations in C. Geometric interpretations.		
2 The Cauchy-Riemann conditions. Holomorphic functions.		
3 Elementary functions, equations in the complex domain.		
4 The complex integral.	Explanation	
5 Series of functions.		
6 Residues. The Residue Theorem.		
7 Computing real integrals by using the Desidue Theorem	Demonstration	
/ Computing real integrals by using the Residue Theorem.	Demonstration	
8 The Fourier transform.	Demonstration Collaboration	
 8 The Fourier transform. 9 Properties and apploications of the Fourier transform 	Demonstration Collaboration	
7 Computing real integrals by using the Residue Theorem. 8 The Fourier transform. 9 Properties and apploications of the Fourier transform 10 The Laplace transform.	Demonstration Collaboration Interactive	
7 Computing real integrals by using the Residue Theorem. 8 The Fourier transform. 9 Properties and apploications of the Fourier transform 10 The Laplace transform. 11 The inverse Laplace transform.	Demonstration Collaboration Interactive activities	
 Computing real integrals by using the Residue Theorem. 8 The Fourier transform. 9 Properties and apploications of the Fourier transform 10 The Laplace transform. 11 The inverse Laplace transform. 12 Applications of the Laplace transform. 	Demonstration Collaboration Interactive activities	
7 Computing real integrals by using the Residue Theorem. 8 The Fourier transform. 9 Properties and apploications of the Fourier transform 10 The Laplace transform. 11 The inverse Laplace transform. 12 Applications of the Laplace transform. 13 The z transform.	Demonstration Collaboration Interactive activities	
7 Computing real integrals by using the Residue Theorem. 8 The Fourier transform. 9 Properties and apploications of the Fourier transform 10 The Laplace transform. 11 The inverse Laplace transform. 12 Applications of the Laplace transform. 13 The z transform. 14 Difference equations solved with the z transform.	Demonstration Collaboration Interactive activities	

1. A.I. Mitrea, Analiza matematica in complex (curs+culegere de probleme), Ed. Mediamira, Cluj-Napoca, 2005.

2. A.I. Mitrea, Transformari integrale si discrete (curs + culegere de probleme) Ed. Mediamira, Cluj-Napoca, 2004.

3. M.L. Krasnov, A.I. Kiselev, G.I. Makarenko, Functions of a Complex Variable, Operational Calculus and Stability Theory, Mir Publishers, Moscow, 1984.

9. Bridging course contents with the expectations of the representatives of the community, professional associations and employers in the field

Collaboration with engineers in order to identify and solve problems raised by the market.

10. Evaluation

Activity type	10.1	Assessment criteria	10.2	Assessment methods	10.3	Weight in the final grade		
Course		Abilities of understanding and		Written examination		30%		
		using creatively the concepts and						
		proofs						
Applications		Abilities of solving problems and		Written examination		70%		
	applying algorithms							
10.4 Minimum standard of performance								
Ability to present coherently a theoretical subject and to solve problems with practical content.								

Course responsible Prof. dr. Ioan Rașa Head of department Prof.dr.eng. Rodica Potolea

1. Data about the program of study

	2 au acout ne program or staal	
1.1	Institution	The Technical University of Cluj-Napoca
1.2	Faculty	Automation and Computer Science
1.3	Department	Computer Science
1.4	Field of study	Computer Science and Information Technology
1.5	Cycle of study	Bachelor of Science
1.6	Program of study/Qualification	Computer Science/ Engineer
1.7	Form of education	Full time
1.8	Subject code	11.

2. Data about the subject

2.1	.1 Subject name				Elect	Electrotechnics					
2.2	Subject area			Com	Computer Science and Information Technology						
2.3	2.3 Course responsible/lecturer					Asso	Assoc. prof. dr. eng. Laura DARABANT – Laura.Darabant@et.utcluj.ro				
2.4	Teachers in charge of applications				As. drd. eng. Mihaela CRETU - Mihaela.Cretu@et.utcluj.ro;						
					As.drd. eng. Denisa STET – Denisa.Stet@et.utcluj.ro						
2.5	Year of study	Ι	2.6	Semester	2	2.7	Assessment	exam	2.8	Subject category	DID/OB

3. Estimated total time

Sem.	Subject name	Lecture	Apj	olicat	ions	Lecture	App	licati	ions	Individual study	TOTAL	Credit
		[hours / week.]		[hours / semester]			ter]					
			S	L	Р		S	L	Р			
2	Electrotechnics	3	-	1	-	42	-	14	-	74	130	5

3.1	Number of hours per week	4	3.2	of which, course	3	3.3	applications	1
3.4	Total hours in the teaching plan	56	3.5	of which, course	42	3.6	applications	14
Individual study								
Manual, lecture material and notes, bibliography								23
Supplementary study in the library, online and in the field								12
Preparation for seminars/laboratory works, homework, reports, portfolios, essays								25
Tuto	ring							10
Exam	s and tests							4
Other activities								
3.7	Total hours of individual study		74					
3.8	Total hours per semester		130					

3.9	Number of credit points	5	

	4. Pre-requisites (where appropria	te)
4.1	Curriculum	
4.2	Competence	Mathematics I, II; Physics

5. Requirements (where appropriate)							
5.1	For the course						
5.2	For the applications	The presence of the lab is mandatory					

	C1 – Operating with basic Mathematical, Engineering and Computer Science concepts
_ s	C1.1 – Recognizing and describing concepts that are specific to the fields of calculability, complexity,
na	programming paradigms, and modeling computational and communication systems
sio	C1.3 – Building models for various components of computing systems
fes	C1.4 – Formal evaluation of the functional and non-functional characteristics of computing systems
Pro	C1.5 – Providing a theoretical background for the characteristics of the designed systems
0	

	N/A	
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11 2 1	/ Discipline Sejeen/es (us results nom the wey competences gamea)					
7.1	General objective					
7.2	Specific objectives					

8.1. I	ecture (syllabus)	Teaching methods	Notes
1	Electric and magnetic quantities. Static electric and magnetic fields (the electric		10005
	field in free space and in material, electric current, the magnetic field in free		
	space and in material)	Multimedia,	
2	Laws and theorems of electromagnetic field	PowerPoint	
3	Electrical capacitance, energy and forces	Presentations,	
4	Magnetic circuits. Self-inductance and mutual inductance. Magnetic energy and forces.	Demonstration board	
5	Basic concepts, units and laws of circuit theory (characteristic values, power in sinusoidal regime, representation of sinusoidal functions by vectors and complex numbers)		
6	The characterisation of the linear circuits in complex plane, the complex form of some theorems		
7	Equivalent impedances (series and parallel connection, without mutual inductance, with mutual inductance, real condenser, real inductance, air core transformer)		
8	Resonance (in series, parallel, real, inductively coupled circuits, power factor improvement)		
9	Two-port networks (equations, equivalent circuits, open-circuit and short-circuit tests, characteristic impedance, propagation constant, filters)		
10	Network theorems (th superposition theorem, Thevenin-Norton theorem, mesh or loop analysis, node analysis, matrix methods)		
11	Transient regime of linear circuits (continuity conditions, transient behaviour of the R-L, R-C and R,L,C)		
12	Transient regime of linear circuits (the Laplace transform, Duhamel integral, state variable method)		
13	Study-state periodic non-sinusoidal regime (Fourier expansion, power, network analysis)		
14	Transmission lines (the primary line parameters, the equations of the transmission line, voltage and current waves on long lines, distortionless lines)		
Biblic	 Dgraphy 1. The Theory of Electric Circuits, authors: RV Ciupa, V. Ţopa, Casa Cartii de ISBN 973-9204-98-8 	Stiinta Publishing H	ouse, 2003
	 Simion, E., Maghiar, T., <i>Electrotehnica</i>, E.D.P., Bucureşti, 1982 Mocanu, C. I., <i>Teoria câmpului electromagnetic</i>, E.D.P., Bucureşti, 1981 		

-	in the carry and the comparison of the carry and the carry		
8.2. A	Applications (Laboratory)	Teaching methods	Notes
1	Determination of the spectrum and equipotential surfaces of an electric field		
	using a electrokinetic model		
2	The study of a magnetic circuit. The measurement of the iron losses using an		
	oscilloscope		
3	Representation of sinusoidal functions by vectors and complex numbers	Practical exercises	
4	Analysis of the R,L,C series and parallel circuits, of the voltage and current		
	resonances		
5	Power transfer in inductively coupled circuits		
6	The study of a circuit in non-sinusoidal regime		
7	The study of the transient regime, methods for solving circuits in transient regime		
Biblic	ography		

- 1. Răduleț, R., Bazele electrotehnicii. Probleme., E.D.P., București, 1981
- 2. Micu, D.D., Creț, Laura, Duma, Denisa, Teoria circuitelor electrice. Culegere de probleme., UTPress, Cluj-Napoca, 2005

9. Bridging course contents with the expectations of the representatives of the community, professional associations and employers in the field

10. Evaluation									
Activity type	10.1	Assessment criteria	10.2	Assessment methods	10.3	Weight in the final grade			
Course				Three hours written		0.8 WT			
				examination, written					
				test (WT)					
Applications				Laboratory works		0.2 LW			
				(LW)					
10.4 Minimum standard of performance									
N=0,8 WT + 0,2 LW									
Pass conditions	Pass conditions: N>5: LW>5								

Course responsible Assoc.prof.dr.eng.Laura Darabant Head of department Prof.dr.ing. Rodica Potolea

1. Data about the program of study

	1 8	
1.1	Institution	The Technical University of Cluj-Napoca
1.2	Faculty	Automation and Computer Science
1.3	Department	Computer Science
1.4	Field of study	Computer Science and Information Technology
1.5	Cycle of study	Bachelor of Science
1.6	Program of study/Qualification	Computer Science/ Engineer
1.7	Form of education	Full time
1.8	Subject code	12.

2. Data about the subject

2.1	.1 Subject name			Digit	Digital Systems Design						
2.2	2.2 Subject area			Com	Computer Science and Information Technology						
2.3	2.3 Course responsible/lecturer			Prof.	Prof. dr. eng. Cret Octavian Augustin – Octavian.Cret@cs.utcluj.ro						
2.4	2.4 Teachers in charge of applications			Dipl. eng. Lorena Dăian – <u>lorenaiulia@yahoo.com</u>							
					Dipl. eng. Bogdan Popa – <u>bogdititupopa@gmail.com</u>						
2.5	Year of study	Ι	2.6	Semester	2	2.7	Assessment	exam	2.8	Subject category	DID/OB

3. Estimated total time

Sem.	Subject name	Lecture	Арр	olicat	ions	Lecture	e Applications		Individual study	TOTAL	Credit	
		[hours / week.]			[hours / semester]							
			S	L	Р		S	L	Р			
2	Digital Systems Design	2	-	2	-	28	-	28	-	74	130	5

3.1	Number of hours per week	4	3.2	of which, course	2	3.3	applications	2
3.4	Total hours in the teaching plan	56	3.5	of which, course	28	3.6	applications	28
Indiv	idual study							Hours
Manual, lecture material and notes, bibliography						25		
Supp	lementary study in the library, online and	l in the fie	ld					17
Prepa	aration for seminars/laboratory works, ho	mework, 1	reports	s, portfolios, essays				17
Tutoring					6			
Exam	s and tests							9
Othe	r activities							0
3.7	Total hours of individual study		74					
3.8	Total hours per semester		130					
3.9	Number of credit points		5					

3.8	Total hours per semester	1
3.9	Number of credit points	

4. Pre-requisites (where appropriate)

4.1	Curriculum	Logic Design
4.2	Competence	At least one high level programming language (i.e. C or PASCAL)

	5. Requirements (where appropriate)						
5.1	For the course	• A minimum of 75% course attendance rate is mandatory for being admitted to					
		the final exam					
5.2	For the applications	• Preliminary preparation of summaries from the indicated bibliography					
		(laboratory textbook)					

	C2 – Designing hardware, software and communication components
s I	C2.1 - Describing the structure and functioning of computational, communication and software components and
ona nce	systems
ssic	C2.2 – Explaining the role, interaction and functioning of hardware, software and communication components
Profes compe	C2.3 – Building the hardware and software components of some computing systems using algorithms, design
	methods, protocols, languages, data structures, and technologies
	C2.4 – Evaluating the functional and non-functional characteristics of the computing systems using specific

	metrics
	C2.5 – Implementing hardware, software and communication systems
	N/A
s nces	
ross eter	
C	
ర	

7.1	General objective	• The main objective of this discipline is to give to the students the bases of Digital Systems Design, in order to make them able to analyze, design and implement any complex digital system.
7.2	Specific objectives	 To reach this goal, students will learn to: Apply Digital System Design principles and descriptive techniques; Understand various aspects of Automata Theory with applications in the field of Digital Systems Design; Describe any digital system in VHDL; Utilize programmable devices such as FPGAs and PLDs to implement digital systems.

8. Contents

8.1. L	ecture (syllabus)	Teaching methods	Notes
1	VHDL hardware description language – basic design units, signals		
2	VHDL hardware description language – generics, constants, operators, data		
	types, attributes		
3	VHDL hardware description language – sequential domain		
4	VHDL hardware description language – concurrent domain		
5	Creating testbenches for simulating and testing circuits in VHDL		
6	Automata (Finite State Machines) Theory - classification, definitions, formal	Dla alth a and	
	models	presentation	N/A
7	Microprogramming	discussions	11/11
8	Microprogrammed Devices	uiscussions	
9	Designing Synchronous Automata		
10	Analysis and Design (Synthesis) of Asynchronous Automata (I)		
11	Analysis and Design (Synthesis) of Asynchronous Automata (II)		
12	Automata Identification		
13	Lossless Machines		
14	Linear Automata		
Biblio	graphy		
1. Dis	zital Design Principles and Practices, John F. Wakerly, Prentice-Hall, 2000.		

Digital Design Principles and Practices, John F. Wakerly, Prentice-Hall, 20
 Automate programabile, Th. Borangiu, R. Dobrescu, Ed. Academiei, 1986.

3. Advanced Digital Logic Design Using VHDL, State Machines, and Synthesis for FPGA's, Sunggu Lee, Thomson-Engineering; 1 edition (April 25, 2005), ISBN 0534466028.

4. PowerPoint slides for VHDL and Automata Theory lectures + sets of problems for the individual study:

http://users.utcluj.ro/~lucia/index.html

8.2. A	Applications (Laboratory)	Teaching methods	Notes
1	Introduction to VHDL	Practical work on	
2	Basic design units in VHDL	test boards, FPGA	N/A
3	Signals, generics, constants, in VHDL	boards,	
4	Operators, data types in VHDL	specialized	
5	Attributes in VHDL	software,	
6	Sequential domain. Processes in VHDL	blackboard	

7	Sequential statements in VHDL	presentations,	
8	Concurrent domain in VHDL	supplemental	
9	Concurrent statements in VHDL	explanations and	
10	Sub-programs in VHDL	discussions	
11	Testbenches in VHDL		
12	Standard and predefined packages in VHDL		
13	Mini-projects delivery		
14	Lab test		
Biblic	ography		

- 1. Limbajul VHDL, Îndrumător de laborator, Ediția a-3-a. O. Creț, L. Văcariu, Ed. U.T. Press, Cluj-Napoca, 2007.
- PowerPoint slides for VHDL and Automata Theory lectures + sets of problems for the individual study: 2. http://users.utcluj.ro/~lucia/index.html

9. Bridging course contents with the expectations of the representatives of the community, professional associations and employers in the field

• Since this discipline is a basic one in Computer Science, its content is "classic" but also modern because it familiarizes students with the modern principles of Logic Design (utilization of modern simulation and synthesis tools, FPGA and CPLD-based design etc.). Its contents have been discussed with major academia and industry actors from Romania, Europe and U.S.A. and it has been evaluated several times by Romanian Governmental Agencies like CNEAA and ARACIS.

10. Evaluation

Activity type	10.1	Assessment criteria	10.2	Assessment methods	10.3	Weight in the final grade				
Course		Problems solving abilities		Written Exam		60%				
Course		Presence, (Inter)activity								
Homeworks		Problems solving abilities		Practical Evaluation		20%				
		Problems solving abilities		Practical Evaluation		20%				
Applications				(hands-on)						
		Presence, (Inter)activity								
10.4 Minimum	stan	dard of performance								
Modeling and	Modeling and solving typical Digital Systems Design problems using the domain-specific formal apparatus									

Course responsible Prof. dr. eng. Creţ Octavian Augustin Head of department Prof.dr.ing. Rodica Potolea

1. Data about the program of study

1.1	Institution	The Technical University of Cluj-Napoca
1.2	Faculty	Automation and Computer Science
1.3	Department	Computer Science
1.4	Field of study	Computer Science and Information Technology
1.5	Cycle of study	Bachelor of Science
1.6	Program of study/Qualification	Computer Science/ Engineer
1.7	Form of education	Full time
1.8	Subject code	13.

2. Data about the subject

2.1	Subject name				Data	Data Structures and Algorithms					
2.2	Subject area			Com	Computer Science and Information Technology						
2.3	2.3 Course responsible/lecturer					S.1. d	S.I. dr. eng. Marius Joldos – Marius Joldos @cs.utcluj.ro				
2.4	Teachers in charge of applications					S.L.dr.mat. Iulia Costin – Iulia.Costin@cs.utcluj.ro					
						As.d	r. eng. Andrei V	√ătavu – Andre	i.Vata	vu@cs.utcluj.ro	
2.5	Year of study	Ι	2.6	Semester	2	2.7	Assessment	exam	2.8	Subject category	DID/OB
	-										

3. Estimated total time

Sem.	Subject name	Lecture	e Applications		Lecture	re Applications		Individual study	TOTAL	Credit		
		[hours / week.]		[hours / semester]			ter]					
			S	L	Р		S	L	Р			
2	Data Structures and Algorithms	2	-	2	-	28	•	28	-	74	130	5

3.1	Number of hours per week	4	3.2	of which, course	2	3.3	applications	2
3.4	Total hours in the teaching plan	56	3.5	of which, course	28	3.6	applications	28
Individual study							Hours	
Manual, lecture material and notes, bibliography							27	
Supplementary study in the library, online and in the field							5	
Preparation for seminars/laboratory works, homework, reports, portfolios, essays							10	
Tuto	ring							7
Exam	s and tests							5
Other activities						0		
3.7	Total hours of individual study		74					
3.8	Total hours per semester		130					

3.8	rotar nours per semester	150	
3.9	Number of credit points	5	

2	4. Pre-requisites (where appropriate)									
4.1	Curriculum	Computer Programming course								
4.2	Competence	Programming in C								

4	5. Requirements (where appropria	ite)
5.1	For the course	
5.2	For the applications	

	C1 – Operating with basic Mathematical, Engineering and Computer Science concepts
	C1.1 – Recognizing and describing concepts that are specific to the fields of calculability, complexity,
na	programming paradigms, and modeling computational and communication systems
sio	C1.2 – Using specific theories and tools (algorithms, schemes, models, protocols, etc.) for explaining the
fes	structure and the functioning of hardware, software and communication systems
Pro	C1.4 – Formal evaluation of the functional and non-functional characteristics of computing systems
- 0	C1.5 – Providing a theoretical background for the characteristics of the designed systems

s	N/A		
s nce			
ros			
S			

7.1	General objective	To acquaint the students with a wide range of fundamental algorithms and data structures. To learn how to use general methods for development of algorithms, as well as mathematical tools for analyzing the correctness and efficiency of algorithms.
7.2	Specific objectives	• To choose the appropriate data structure for modelling a given problem.
		• To compare and contrast the cost and benefits of dynamic and static structure implementations.
		• To compare iterative and recursive solutions for elementary problems.
		• To determine when a recursive solution is appropriate for a problem.
		• To determine the time and space complexity of simple algorithms and recursively defined algorithms.
		• To design and implement algorithms using development techniques such as: greedy, divide-and-conquer, backtracking, dynamic programming, branch and bound.
		• To write C programs that use data structures such as: arrays, linked lists, stacks, queues, trees, hash tables, and graphs.
		• To implement in C the most common sorting algorithms.
		• To solve problems using the fundamental graph algorithms, including depth-first and breadth-first search, topological sort, minimum spanning tree algorithm, and single-source shortest path.

8. Contents

0.1 1		T 1: (1 1	NT /
8.1. Le	ecture (syllabus)	Teaching methods	Notes
1	Introduction. Problem solving. Measuring time efficiency of algorithms – Big-Oh notation.		
	Stack, Queue, List ADTs.		
2	Trees – definitions, traversals. ADT Tree. Implementations. Binary Search Trees. Optimal		
	Trees		
3	Sets ADTs and Implementations. Dictionary ADT. Hash Tables. Mapping ADT. Priority		
	Queue ADT.		
4	Advanced Set Representation Methods. AVL trees. 2-3 Trees. Union-Find Set ADT.		
5	Directed Graphs. Definitions. Representations. ADT's. Single Source Shortest Path		
	Problem (Dijkstra, Bellman-Ford, Floyd-Warshall). Traversals for DGs. Parenthesis		
	Lemma. DAGs. Strong Components. Topological Sort		
6	Undirected Graphs. Terminology. Free Trees. Graph Representations. Minimum Spanning	Lectures, demos	Uses a
	Trees (algorithms: Prim, Kruskal). Graph Traversals (depth-first, breadth-first).	and discussions	video-
	Articulation points & Biconnected Components. Graph Matching.		projector
7	Algorithm Analysis. Correctness of Algorithms. Efficiency of Algorithms		
8	Algorithm Design techniques I. Divide-and-Conquer. Dynamic Programming		
9	Algorithm Design techniques II. Brute Force Algorithms. Greedy Algorithms.		
	Backtracking		
10	Algorithm Design techniques III. Minimax. Alpha-Beta Prunning. Search Tree Strategies		
	(backtracking revisited, branch and bound). Local Search.		
11	Sorting. Simple comparison sorting schemes (bubble, selection, insertion). HeapSort.		
	QuickSort. Decision Tree model. Counting Sort, Radix Sort, Bucket Sort. Criteria for		
	Sorting Algorithm Selection.		
12	Data Structures and Algorithms for External Storage I. External Sorting BTrees		
13	Data Structures and Algorithms for External Storage I. B+Trees		
14	Review		
Biblio	graphy		

Aho, Hopcroft, Ullman. Data Structures and Algorithms, Addison-Wesley, 427 pages, 1987.
 Cormen, Leiserson, Rivest, Stein: Introduction to Algorithms, 2nd edition. MIT Press / McGraw Hill, 1028 pages,

2001.

3. Preiss, Bruno. Data Structures and Algorithms with object-Oriented Design Patterns in C++, John Wiley and Sons, 660 pages, 1999 (freely available on the Web)

page	s, is the second s				
8.2. A	Applications (Laboratory)	Teaching methods	Notes		
1	Singly Linked Lists. Stacks. Queues				
2	Circular Lists. Circular Queues				
3	Doubly Linked lists				
4	Arbitrary Trees		PCs		
5	Binary Search Trees		equipped		
6	Hash Tables	Tutoring,	with		
7	Graph Representations and Traversals	discussions, and	MinGW		
8	Graph Processing Algorithms	assisted program	C and		
9	Algorithm Design I. Greedy and Backtrack	development	Code-		
10	Algorithm Design II. Divide & Conquer and Branch and Bound blocks				
11	Algorithm Design III. Dynamic Programming and Heuristics.		IDE		
12	Fundamental Sorting Algorithms				
13	Comparison of Algorithm Performance (estimated vs. practical) I				
14	Comparison of Algorithm Performance (estimated vs. practical) II				
Biblic	ography				
1	. Moodle course Web Site available at https://193.226.5.110				

9. Bridging course contents with the expectations of the representatives of the community, professional associations and employers in the field

10. Evaluation

Activity type	10.1	Assessment criteria	10.2	Assessment methods	10.3	Weight in the final grade
Course		The understanding of the concepts		Written exam		60%
		taught and the ability to solve				
		problems				
Applications		Quality of the assigned applications		Analysis and		40%
				evaluation of the		
				solved assignments		
10.4 Minimum standard of performance						
Correct solutions for min 60% of the exam tonics and applications						

Course responsible S.l. dr. eng. Marius Joldoş Head of department Prof.dr.ing. Rodica Potolea

1. Data about the program of study

-	Duta about the program of study	
1.1	Institution	The Technical University of Cluj-Napoca
1.2	Faculty	Automation and Computer Science
1.3	Department	Computer Science
1.4	Field of study	Computer Science and Information Technology
1.5	Cycle of study	Bachelor of Science
1.6	Program of study/Qualification	Computer Science/ Engineer
1.7	Form of education	Full time
1.8	Subject code	14.

2. Data about the subject

2.1	Subject name Foreign Language II (English, French, German)										
2.2	Subject area Computer Science and Information Technology										
2.3	Course response	ible/l	lectui	rer							
2.4	Teachers in cha	arge o	of app	olications		Asist	.drd. Ema Ada	ım, <u>adam@lang</u>	.utclu	<u>j.ro</u>	
						Asist	.drd. Monica N	legoescu, <u>Nego</u>	escu (<u>@mail.utcluj.ro</u>	
						Asist	.dr. Sanda Pădu	ırețu <u>Sanda.Pad</u>	luretu	<u>@lang.utcluj.ro</u>	
						Asist.dr. Maria Olt maria.olt@lang.utcluj.ro					
						Asist.dr. Cecilia Policsek <u>cecilia.policsek@lang.utcluj.ro</u>					
						Asist.dr. Florina Codreanu <u>codreanu.florina@gmail.com</u>					
							dr. Mona Trip	on <u>Mona.Tripor</u>	n@lar	<u>ig.utcluj.ro</u>	
	Asist. drd. Aurel Bărbînță Aurel. Barbinta@lang.utcluj.ro										
	Asist.dr. Adina Forna adina.forna@yahoo.com										
2.5	Year of study	Ι	2.6	Semester	2	2.7	Assessment	Colloquium	2.8	Subject category	DC/OB
								-			

3. Estimated total time

Sem.	Subject name	Lecture	App	olicat	ions	Lecture	App	licati	ions	Individual study	TOTAL	Credit
		[hou	rs / v	veek.]	[hour	s / se	emes	ter]		
			S	L	Р		S	L	Р			
2	Foreign Language II (English, French, German)	-	2	-	-	-	28	-	-	24	52	2

3.1	Number of hours per week	2	3.2	of which, course	-	3.3	applications	2
3.4	Total hours in the teaching plan	28	3.5	of which, course	-	3.6	applications	28
Individual study								Hours
Manual, lecture material and notes, bibliography							8	
Supp	lementary study in the library, online and	l in the fie	ld					4
Preparation for seminars/laboratory works, homework, reports, portfolios, essays								8
Tutoring								
Exams and tests							4	
Othe	r activities							
3.7	Total hours of individual study		24					
3.8	Total hours per semester		52	1				

- L		"	_
	3.9	Number of credit points	

4. Pre-requisites (where appropriate)

4.1	Curriculum	A2/B1 according to the Common European Framework for Languages
4.2	Competence	Team work

2

5. Requirements (where appropriate)

5.1	For the course	N/A
5.2	For the applications	Seminar attendance compulsory

	N/A
Professional Competences	
Cross competences	CT2 – Identifying, describing and conducting processes in the projects management field, assuming different roles inside the team and clearly and concisely describing, verbally or in writing, in Romanian and in an international language, the own results from the activity field

7.1	General objective	Development of communicative competence in an engineering professional context
7.2	Specific objectives	 Mastering basic vocabulary and language structures typical of sciences studied Development of the skill of writing short technical texts and of presenting them

8. Cor	ntents		
8.1. L	ecture (syllabus)	Teaching methods	Notes
1			
Biblio	graphy		
8.2. A	Applications (Seminars)	Teaching methods	Notes
1	Engineering and automation.		
2	Microelectronics and nanotechnology		
3	Computers in industry		
4	Design of products. Definition		
5	Procedures	Conversation,	
6	Systems of communication	improving the	
7	Monitoring	speaking, writing,	
8	Types of networks, The Internet	listoning skills	
9	Engineers and managers	working in pairs	
10	The responsibilities of the manager	and groups	
11	Companies	and groups	
12	Organisations and their culture		
13	Final test]	
14	Final test]	
D'1 1'	1		

Bibliography

- 1. Munteanu, S-C. (2004) Reading skills For Engineering Students, UTPress, Cluj-Napoca.
- 2. Granescu, M. et. al. Students' Grammar Of English, UTPress, Cluj-Napoca, 2001.
- 3. Bonamy, D. Technical English 1-2, Longman, London
- 4. Tripon, Mona: Faszination Technik. Sprachtrainer Deutsch für Studenten technischer Universitäten. Editura Napoca Star, Cluj-Napoca, 2012. ISBN 978-973-647908-3
- 5. Odou M., Informatique.com, Clé international, 2010
- 6. Constantin Paun, Limba franceză pentru știință și tehnică, EdituraNiculescu, Bucuresti, 1999
- 7. Vlaicu, R., Grammaire du français scientifique et technique, Cluj-Napoca, UTPRESS, ISBN 2007 973-662-2258-4.

9. Bridging course contents with the expectations of the representatives of the community, professional associations and employers in the field

Mastering a foreign language will help students in a more flexible integration in the labour market, and have improved personal development. The introduction in the language for specific purposes will facilitate reading more documents in the field of study.

10. Evaluation							
Activity type	10.1	Assessment criteria	10.2	Assessment methods	10.3	Weight in the final grade	
Course							
ApplicationsAssignments and tests are corrected and marked if submitted in due time. The undergraduate will be allowed to sit in the final test if he/she attends seminars in a proportion of 80% of the time			Written test, Oral test		100%.		
10.4 Minimum	stan	dard of performance		·			
The undergradu	iate w	vill be allowed to sit in the final test if h	e/she	attends seminars in a pro	oporti	on of 80% of the time.	
Final score: attendance= 1pct, written test=5 pct, oral test=4 pct.							
Pass score is re-	ceived	a if 60 % of both tests is produced by th	ne und	ergraduate.			

Head of department Prof.dr.eng. Rodica Potolea Course responsible Conf.univ.dr. Marinela Grănescu Teachers in charge of applications

Asist. drd. Ema Adam, Asist.drd. Monica Negoescu, Asist.dr. Sanda Pădurețu Asist.dr. Maria Olt Asist.dr. Cecilia Policsek Lect. dr. Mona Tripon Asist. drd. Aurel Bărbînță Asist. dr. Forina Codreanu Asist. dr. Adina Forna

1. Data about the program of study

	i Duiu ucour ine program or staalj	
1.1	Institution	The Technical University of Cluj-Napoca
1.2	Faculty	Automation and Computer Science
1.3	Department	Computer Science
1.4	Field of study	Computer Science and Information Technology
1.5	Cycle of study	Bachelor of Science
1.6	Program of study/Qualification	Computer Science/ Engineer
1.7	Form of education	Full time
1.8	Subject code	15.

2. Data about the subject

2.1	Subject name S				Sport	Sport II					
2.2	2.2 Subject area			Com	Computer Science and Information Technology						
2.3	2.3 Course responsible/lecturer			Asso	Assoc.prof. Marin Dumitrescu, PhD, marind@efs.utcluj.ro						
2.4	.4 Teachers in charge of applications				Assoc.prof.Viorel Moisin, PhD, Lecturer Alina Rusu, PhD, Lecturer						
	Mihai Olanescu, PhD student, As. Prof. Bogdan Tanase										
2.5	Year of study	Ι	2.6	Semester	2	2.7	Assessment	verification	2.8	Subject category	DC/OB

3. Estimated total time

Sem.	Subject name	Lecture	Арр	Applications Lecture Applic		licat	ions	Individual study	TOTAL	Credit		
		[hours / week.]			[hours / semester]							
			S	L	Р		S	L	Р			
2	Sport II	-	2	-	-	-	28	•	-	-	28	1

3.1	Number of hours per week	2	3.2	of which, course	-	3.3	applications	2
3.4	Total hours in the teaching plan	28	3.5	of which, course	-	3.6	applications	28
Individual study								
Manual, lecture material and notes, bibliography								
Supplementary study in the library, online and in the field								
Prepa	aration for seminars/laboratory works, ho	mework, 1	reports	, portfolios, essays				
Tuto	ring							
Exam	is and tests							
Other activities								
3.7	Total hours of individual study		-					
3.8	3.8 Total hours per semester 28							

4. Pre-requisites (where appropriate)

3.9 Number of credit points 1

4.1	Curriculum	
4.2	Competence	physically fit, necessary skills, knowledge, skills and abilities gained in classes I-XII

5. Requirements (where appropriate)

5.1	For the course	Muncii Blvd, no.103-105, Cluj-Napoca, Politehnica Swimming Complex
5.2	For the applications	Sports Hall, Muncii Blvd, no.103-105, Cluj-Napoca
		Outdoor and Fitness - Complex Polytechnic

	N/A				
al es					
on					
ssi ete					
np					
Prc					
J					
ЧЗ					

Cross	competences
-------	-------------

CT2 – Identifying, describing and conducting processes in the projects management field, assuming different roles inside the team and clearly and concisely describing, verbally or in writing, in Romanian and in an international language, the own results from the activity field.

7. Discipline objectives (as results from the *key competences gained*)

	1 5	
7.1	General objective	Harmonious physical development
		Maintain health at a high standard
7.2	Specific objectives	Capacity development effort
		 Learning and motor skills development
		Education volitional qualities

8. Contents

8.1. Lect	ure (syllabus)	Teaching methods	Notes			
1						
Bibliogra	phy					
8.2. Apj	plications (Seminars)	Teaching methods	Notes			
1-2	Improvement and maintenance of health, athletic ability and fitness					
3-4	Improving tehnical exercises learned before using tactic tasks					
5-6	Automatization of technical and tactics in game conditions (competition).					
7-8	Learning regulations of different sports, to be able to practice and organize	interactive				
	leisure-time sport activity.	interactive				
9-10	Necessary skills to practice independent physical activity					
11-12	Improving the drills, combinations, schemes in different sport games					
13-14	Close the school situation by passing physical test					
Bibliography						
1. Curs de Educație fizică – Litografiat UTC-N						
2. Dez	2. Dezvoltare fizică generală pentru studenți – UTC-N					
3. Cul	3. Cultură fizică pentru tineret - UTPRES					

9. Bridging course contents with the expectations of the representatives of the community, professional associations and employers in the field

Sports activity there in the curriculum of universities and faculties in the country and abroad. Content is consistent with the expectations of professional associates and employers epistemic community representative of the afferent program.

10. Evaluation

Activity type	10.1	Assessment criteria	10.2	Assessment methods	10.3	Weight in the final grade		
Course		-		-				
Applications		70% + 30% Frequency Active		By passing control				
		Participation, sports skills and		samples				
		advances						
10.4 Minimum standard of performance								

Course responsible Assoc. Prof. Marin Dumitrescu, PhD Head of department Prof.dr.eng. Rodica Potolea

1. Data about the program of study

	2 au acour me program or staal	
1.1	Institution	The Technical University of Cluj-Napoca
1.2	Faculty	Automation and Computer Science
1.3	Department	Computer Science
1.4	Field of study	Computer Science and Information Technology
1.5	Cycle of study	Bachelor of Science
1.6	Program of study/Qualification	Computer Science/ Engineer
1.7	Form of education	Full time
1.8	Subject code	101.

2. Data about the subject

2.1	Subject name				Fund	Fundamentals of Electronic Circuits					
2.2	Subject area			Comp	Computer Science and Information Technology						
2.3	Course responsible/lecturer				Prof.	Prof. Gabriel OLTEAN, PhD					
2.4	Teachers in charge of applications					Assis	Assist.prof. Emilia Şipoş, PhD				
2.5	Year of study	Ι	2.6	Semester	2	2.7	Assessment	Colloquium	2.8	Subject category	DID/FAC
	-							_			

3. Estimated total time

Sem.	Subject name	Lecture	Арј	olicati	ions	Lecture	App	licati	ions	Individual study	TOTAL	Credit
		[hours / week.]		[hours / semester]				ter]				
			S	L	Р		S	L	Р			
2	Fundamentals of Electronic Circuits	2	1	1	-	28	14	14		74	130	5

3.1	Number of hours per week	4	3.2	of which, course	2	3.3	applications	2
3.4	Total hours in the teaching plan	56	3.5	of which, course	28	3.6	applications	28
Individual study								
Man	ual, lecture material and notes, bibliograp	hy						28
Supp	lementary study in the library, online and	in the fiel	ld					12
Prepa	aration for seminars/laboratory works, ho	mework, r	reports	, portfolios, essays				28
Tuto	ring							3
Exam	s and tests							3
Other activities								-
3.7	Total hours of individual study		74					
3.8	Total hours per semester		130					

4. Pre-requisites (where appropriate)

Number of credit points

4.1	Curriculum	
4.2	Competence	Basic knowledge about electrical signals, electric circuits, passive electronic components
		components

5

5. Requirements (where appropriate)					
5.1	For the course	Cluj-Napoca			
5.2	For the applications	Cluj-Napoca			

6. Specific competences

3.9

1	
Professional ompetences	C1 – Operating with basic Mathematical, Engineering and Computer Science concepts
	C1.1 – Recognizing and describing concepts that are specific to the fields of calculability, complexity,
	programming paradigms, and modeling computational and communication systems
	C1.3 – Building models for various components of computing systems
	C1.4 – Formal evaluation of the functional and non-functional characteristics of computing systems
	C1.5 – Providing a theoretical background for the characteristics of the designed systems
- 0	

	N/A	
s nces		
ross etei		
omp C		
CC		

7.1	General objective	Developing the competences regarding the use of electronic devices, regarding the use, analysis and (re)design of fundamental electronic circuits.
7.2	Specific objectives	 Recognizing and understanding basic concepts that are specific to electronic devices, fundamental electronic circuits. Developing skills and abilities necessary for the use of electronic devices in simple electronic circuits Developing skills and abilities necessary for the use of electronic circuits Developing skills and abilities for the analysis and (re)design of electronic circuits.

8. Contents

8.1. L	ecture (syllabus)	Teaching methods	Notes
1	Introduction. Fundamentals: electrical signals, relations and theorems for	Presentation,	Use of .ppt
	electric circuits.	euristic	presentatio
2	Diodes. Models for switching diode. Switching DR circuits. Switching DC	conversation,	n,
	circuits. Single-phase rectifiers with capacitive filter. Zener Diode. LED	exemplification,	projector,
3	Operational amplifier (op amp). Op-amp terminals. Op-amp operation. Ideal op	problem	blackboard
	amp. Modes of use.	presentation,	
4	Simple op-amp comparators. Inverting and noninverting comparators. Voltage	teaching exercise,	
	transfer characteristic. Waveforms	case study,	
5	Positive feedback op-amp comparators. Inverting and noninverting	formative	
	comparators. Voltage transfer characteristic. Waveforms	evaluation	
6	Negative feedback op-amp amplifiers. Inverting, noninverting amplifiers:		
	voltage transfer characteristic, waveforms, gain, input and output resistances.		
7	Op-amp applications: summing amplifiers, differential amplifiers, voltage		
	domain conversion circuits, integrator and differentiator; precision rectifier.		
8	Transistor digital circuits. MOSFET Digital Circuits. Bipolar digital circuits.		
	Noise margins.		
9	DC voltage regulators. Parametric regulators. Linear voltage regulators with op		
	amp. Increasing the output current. Over - current and short - circuit protection.		
10	Integrated voltage regulators. The 723 voltage regulator. Three – terminal fixed		
	regulator. Switching voltage regulators.		
11	Sinusoidal oscillators. Oscillation criterion. RC oscillators. Op - amp and Wien		
	bridge oscillators. Automatic control of the amplitude. Op amp and RC ladder		
	network oscillator.		
12	Nonsinusoidal oscillators. Astable multivibrators. Astable multivibrator with		
	one op – amp. Astable multivibrator with an integrator and a comparator.		
	Quartz – crystal clock generator. LM555 timer.		
13	Power amplifiers. Amplifier classes. Class B amplifiers. Operating principle,		
	VTC, crossover distortions, waveforms, powers, efficiency.		
14	Class AB amplifiers. Biasing using diodes. Biasing using V_{BE} multiplier.		
	Overcurrent protection. Use of compound transistors with higher current gain.		
Biblio	graphy		
1. Olt	ean, G., Electronic Devices, Editura U.T. Pres, Cluj-Napoca, ISBN 973-662-220-7,	2006; 317 pag.	
2. Olt	ean, G., Circuite electronice, UT Pres, Cluj-Napoca, 2007, ISBN 978-973-662-300	-4, 203 pag.	
3. Se	dra, A. S., Smith, K. C., Microelectronic Circuits, Fifth Edition, Oxford University	Press,	
ISBN	: 0-19-514252-7, 2004.		
8.2. A	Applications	Teaching methods	Notes
Semin	nars		
1	Fundamentals		Use of
2	Diodes	Didactic and	laboratory

3	Op-amp comparators	experimental proof,	instruments,
4	Op-amp amplifiers. Logic Circuits with Transistors	didactic exercise,	experimental
5	Voltage Regulators. Integrated Voltage Regulators	team work	boards,
6	Sinusoidal Oscillators. Nonsinusoidal oscillators		computers,
7	Power Amplifiers. Review		magnetic
Labor	ratory		board,
1	Lab instrumentation		blackboard
2	Applications of DR circuits		
3	Op-Amp voltage comparator		
4	Op-Amp basic amplifier		
5	LM 7805 voltage regulator		
6	Class B amplifier		
7	Laboratory test		

Bibliography

1. Oltean, G., Sipos, Emilia, Miron, C., Ivanciu, Laura, Laboratory Manual for Electronic Devices, Editura UTPRESS, Cluj Napoca, 2010, ISBN 978-973-662-542-8, 90 pag.

2. Şipoş, Emilia, Oltean, G., Miron, C., Ivanciu, Laura, Gordan, Mihaela, Fundamental Electronic Circuits. Laboratory Manual, UT Pres, Cluj-Napoca, 2009, ISBN 978-973-662-503-9; 91 pag

On – line references

1. Oltean, G., Fundamentals of Electronic Circuits, PowerPoint slides,

http://www.bel.utcluj.ro/dce/didactic/fec_aai/fec_aai.htm

2. Oltean, G, et al., Fundamentals of Electronic Circuits. Seminars and laboratories,

http://www.bel.utcluj.ro/dce/didactic/fec_aai/fec_aai.htm

9. Bridging course contents with the expectations of the representatives of the community, professional associations and employers in the field

The discipline content and the acquired skills are in agreement with the expectations of the professional organizations and the employers in the field, where the students carry out the internship stages and/or occupy a job, and the expectations of the Romanian Agency for Quality Assurance (ARACIS).

10. Evaluation

Activity type	10.1	Assessment criteria	10.2	Assessment methods	10.3	Weight in the final grade	
Course		The level of theoretical knowledge and practical skills acquired for the analysis and (re)design of electronic circuits		- 3 formative evaluation tests (problem solving)		- T, max 10 pts. 10%	
				- Summative evaluation written exam (theory and problems)		- E, max 10 pts. 60%	
Applications		The level of the abilities acquired for problem solving and experimental analysis of electronic circuits		- Continuous formative evaluation		- L, max. 10 pts. 20% - S, max. 10 pts. 10%	
10.4 Minimum standard of performance							
	$L \ge 5, E \ge 4$ 0,6E+0,1T+0,2L+0,1S ≥ 4.5						

Course responsible Prof. Gabriel OLTEAN, PhD Head of department Prof.dr.eng. Rodica POTOLEA

1. Data about the program of study

	I 8	
1.1	Institution	The Technical University of Cluj-Napoca
1.2	Faculty	Automation and Computer Science
1.3	Department	Computer Science
1.4	Field of study	Computer Science and Information Technology
1.5	Cycle of study	Bachelor of Science
1.6	Program of study/Qualification	Computer Science/ Engineer
1.7	Form of education	Full time
1.8	Subject code	102.

2. Data about the subject

2.1	Subject name				Chemistry						
2.2	2 Subject area				Com	Computer Science and Information Technology					
2.3	Course responsible/lecturer					Assoc. prof. chem. Mihaela-Ligia Unguresan;					
	-					Mihaela.Unguresan@chem.utcluj.ro					
2.4	Teachers in cha	arge o	of app	olications		Assoc.prof. chem. Mihaela-Ligia Unguresan;					
						Miha	ela.Unguresan	<u>@chem.utcluj.re</u>	<u>)</u>		
2.5	Year of study	Ι	2.6	Semester	2	2.7	Assessment	exam	2.8	Subject category	DF/FAC

3. Estimated total time

Sem.	Subject name	Lecture Applications		Lecture	ure Applications			Individual study	TOTAL	Credit		
		[hours / week.]			[[hours / semester]			ter]			
			S	L	Р		S	L	Р			
2	Chemistry	2	•	2	-	28	-	28	•	48	104	4

3.1	Number of hours per week	4	3.2	of which, course	2	3.3	applications	2	
3.4	Total hours in the teaching plan	56	3.5	of which, course	28	3.6	applications	28	
Individual study Hou									
Manual, lecture material and notes, bibliography								14	
Supplementary study in the library, online and in the field								10	
Prepa	ration for seminars/laboratory works, ho	mework, r	reports	, portfolios, essays				10	
Tuto	ring							8	
Exam	s and tests							6	
Other activities								0	
3.7	Total hours of individual study		48						

	5.7	Total hours of hidfviddal study	-0
	3.8	Total hours per semester	104
	3.9	Number of credit points	4
Ĩ			

4. Pre-requisites (where appropriate)

4.1	Curriculum	General knowledge of chemistry in high school
4.2	Competence	Arithmetics, Algebra, Mathematical analysis; Physics.

5. Requirements (where appropriate)

5.1	For the course	-
5.2	For the applications	-

nal Ices	C1 – Operating with basic Mathematical, Engineering and Computer Science concepts
	C1.1 – Recognizing and describing concepts that are specific to the fields of calculability, complexity,
sio	programming paradigms, and modeling computational and communication systems
fes	C1.3 – Building models for various components of computing systems
Pro	C1.5 – Providing a theoretical background for the characteristics of the designed systems
_ 0	

s nces	N/A			
Cross npete				
cor				

	1 5	
7.1	General objective	Throughout the semester, this course will touch on many different aspects in the field of
		chemistry. Each one of you should gain knowledge in the field and better appreciate the
		connection between chemistry and everyday life and more specifically how chemistry is
		relevant to biological processes and the health industry. Upon successful completion of this
		course, students will be able:
		- to classify basic forms of matter;
		- to perform mathematical unit conversions;
		- to describe atomic structure and how it affects the structure of the Periodic Table of
		Elements, apply basic concepts of chemical bonding and predict simple molecular formulas,
		and write and analyze chemical formulas;
		- to know the interest materials in the electro techniques, electronics, communications,
		automation and computers: metals and alloys, plastics and semiconductors;
		- to monitor the automated methods for the implementation of fixing the coefficients of
		chemical reactions;
		- to predict, depict and describe: gas behavior, basic properties of chemical bonding,
		molecular geometry and theory of bonding, liquids and intermolecular forces;
		- to deepen the phenomena of electrolysis, electroplating, cathodic deposition, the
		phenomena of corrosion and corrosion protection.
7.2	Specific objectives	- To know how to use the apparatus and glassware from the chemistry laboratory, how to
		measure temperature, pressure, concentration, titer or the purity of some substances or
		solutions; how to analyze the experimental chemical data obtained.
		- To follow the application of the methods for the establishment of the coefficients of
		chemical reactions. Understand and apply concepts to solve problems using: matter and
		measurement, atoms, molecules and ions, stoichiometry & calculations/chemical formulas
		equations
		- To know how to measure the electrode potential, the pH of a solution of metal.
		After reading discipline students will be able to:
		- analyze the chemical substances in a qualitatively and quantitatively mode;
		- know how to interpret graphical results obtained as a result of the kinetic study of
		chemical reactions, of the thermodynamics of a chemical process.

8. Contents

8.1. Le	ecture (syllabus)	Teaching methods	Notes
1	Fundamental concepts in chemistry.(general presentation; chemistry classification; the distribution of elements in nature, chemical compound, substance quantity)	Lecture by teacher; Class discussion	
2	The periodical system of elements (atom components; radioactivity; periodic system structure; physical and chemical properties)	conducted by teacher; Ppt.	
3	Chemical bonds (ionic bond, covalent polar and unpolar bond; metallic bond; Van der Waals, dipole-dipole, ion-dipole, hydrogen bonds)	Presentation; Tutorials;	
4	The gas state (law gases; real gases; virial coefficients; Van der Waals equation)	Coaching: special assistance	
5	Liquid state. Solid state. (viscosity coefficient; vapor pressure; surface tension; crystalline substances, amorphous solid; crystalline systems; state transformations)	provided for students having difficulty in the	
6	Metals (nonferrous, fusible; precious metals; superconductivity)	course.	
7	Ceramic materials (history; ferromagnetic, ferroelectrics, piezoelectric materials; refractors; radio ceramics)		
8	Semiconductors (quantum mechanics, orbital functions; Schrödinger equation; bands formation; semiconductor combinations; impurification; Schottky and Frenkel defects; integrate circuits)		
9	Thermodynamics concepts (thermodynamic system state; state variables;		

		1	
	thermodynamic equilibrium; first and second laws of thermodynamics and		
1.0	their consequences)	-	
10	Thermochemistry (calorimetry; Lavoisier-Laplace's law, Hess's law; applications)		
11	Chemical equilibrium (masses action law: chemical equilibrium in		
	homogeneous systems: relations between K_n K_n and K_n : heterogeneous		
	chemical equilibrium: dimensions characteristic to chemical equilibrium:		
	annlications)		
12	The kinetics of chemical reactions (reaction rate: order rate (0, 1, 2, 3)	1	
12	fractional): reaction mechanism: kinetic simple reaction and complex		
	(successive parallel opposite with preequilibrium): reaction in chain:		
	(successive, parallel, opposite, with preequinoritani), reaction in chain,		
13	Electrochemistry (electrolytic dissociation: electrodes: notentials of electrodes:	-	
15	electrolysis: Butler-Volmer equation: galvanic cells: accumulators)		
14	Metal corrosion Anticorrosion protection		
14	General terms: influencing factors in the process of corrosion: monitoring		
	methods based on thermodynamic stability of the metal: correspondent of the		
	methods		
Biblio	aranhy		
1 M	graphy [J. Ungurasan, Dalia Maria Cligar, <i>Canaral Chamistry</i> , Ed. UTDDESS, Chui	Nanoca ISBN: 078	073 662 707 1
201	2 ng 400	Inapolea, ISBIN. 976-	975-002-707-1,
201 2 M	2, pg. 490. [.] Unguresan I. Jantschi <i>Thermodinamics and chemical kinetics</i> . Ed Mediami	ira Clui-Nanoca 200	5
2. IVI	Intropy M. L. Januson, <i>Intermolational Chapters of Chamistry for automatics</i> , Ed. 117	ra, Ciuj-Napoca, 200	2. 2002
J. L.	Colosi M Abrudean M I Unguresan V Muresan Numerical Simulation M	athod for Distribute	d Parameters
1. Pro	coosses using the Matrix with Partial Derivatives of the State Vector Ed Spring	er ISBN 978-3-319-0	$10013_8(Print)$
978	$_2$ -3.19-00014-5 (Online) 2013 ng 343		50015 0(1 mit),
110	5 517 00014 5 (Online), 2015, pg. 545.		
82	Applications (Laboratory)	Teaching methods	Notes
8.2. A	Applications (Laboratory) Presentation of work Safety norms Analytical balance, Chemical laboratory	Teaching methods	Notes
8.2. A	Applications (Laboratory) Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils glassware and laboratory equipment	Teaching methods Using and	Notes
8.2. 4	Applications (Laboratory) Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils, glassware and laboratory equipment Acid-base titration. Determination by titration of acetic acid content of vinegar.	Teaching methods Using and organising	Notes
8.2. <i>A</i> 1 2	Applications (Laboratory) Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils, glassware and laboratory equipment Acid-base titration. Determination by titration of acetic acid content of vinegar Determination of molar mass of carbon dioxide	Teaching methods Using and organising techniques,	Notes
8.2. <i>A</i> 1 2 3 4	Applications (Laboratory) Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils, glassware and laboratory equipment Acid-base titration. Determination by titration of acetic acid content of vinegar Determination of molar mass of carbon dioxide Hydrated Ionia Compound	Teaching methods Using and organising techniques, apparatus and	Notes
8.2. <i>A</i> 1 2 3 4 5	Applications (Laboratory) Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils, glassware and laboratory equipment Acid-base titration. Determination by titration of acetic acid content of vinegar Determination of molar mass of carbon dioxide Hydrated Ionic Compound	Teaching methods Using and organising techniques, apparatus and materials;	Notes
8.2. <i>A</i> 1 2 3 4 5 6	Applications (Laboratory) Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils, glassware and laboratory equipment Acid-base titration. Determination by titration of acetic acid content of vinegar Determination of molar mass of carbon dioxide Hydrated Ionic Compound Caffeine isolation	Teaching methods Using and organising techniques, apparatus and materials; Observing,	Notes Mathematic al modeling
8.2. <i>A</i> 1 2 3 4 5 6 7	Applications (Laboratory) Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils, glassware and laboratory equipment Acid-base titration. Determination by titration of acetic acid content of vinegar Determination of molar mass of carbon dioxide Hydrated Ionic Compound Caffeine isolation Hydrolyze Determination for the lame to the lame to life	Teaching methods Using and organising techniques, apparatus and materials; Observing, measuring and	Notes Mathematic al modeling and
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8.2. <i>A</i> 1 2 3 4 5 6 7 8 9 1	Applications (Laboratory) Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils, glassware and laboratory equipment Acid-base titration. Determination by titration of acetic acid content of vinegar Determination of molar mass of carbon dioxide Hydrated Ionic Compound Caffeine isolation Hydrolyze Determination of enthalpy, entropy and free enthalpy at different temperatures The heat of hydration of copper sulfate Thermal analysis	Teaching methods Using and organising techniques, apparatus and materials; Observing, measuring and recording; Handling experimental	Notes Mathematic al modeling and numerical simulations, experimenta
8.2. <i>A</i> 1 2 3 4 5 6 7 8 9 10	Applications (Laboratory) Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils, glassware and laboratory equipment Acid-base titration. Determination by titration of acetic acid content of vinegar Determination of molar mass of carbon dioxide Hydrated Ionic Compound Caffeine isolation Hydrolyze Determination of enthalpy, entropy and free enthalpy at different temperatures The heat of hydration of copper sulfate Thermal analysis Acidity of solutions. Conductivity measurement	Teaching methods Using and organising techniques, apparatus and materials; Observing, measuring and recording; Handling experimental observations and	Notes Mathematic al modeling and numerical simulations, experimenta l apparatus.
8.2. <i>A</i> 1 2 3 4 5 6 7 8 9 10 11	Applications (Laboratory) Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils, glassware and laboratory equipment Acid-base titration. Determination by titration of acetic acid content of vinegar Determination of molar mass of carbon dioxide Hydrated Ionic Compound Caffeine isolation Hydrolyze Determination of enthalpy, entropy and free enthalpy at different temperatures The heat of hydration of copper sulfate Thermal analysis Acidity of solutions. Conductivity measurement Reaction rate. The kinetic of simple and complex reactions	Teaching methods Using and organising techniques, apparatus and materials; Observing, measuring and recording; Handling experimental observations and data; Planning	Notes Mathematic al modeling and numerical simulations, experimenta l apparatus.
8.2. <i>A</i> 1 2 3 4 5 6 7 8 9 10 11 12	Applications (Laboratory) Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils, glassware and laboratory equipment Acid-base titration. Determination by titration of acetic acid content of vinegar Determination of molar mass of carbon dioxide Hydrated Ionic Compound Caffeine isolation Hydrolyze Determination of enthalpy, entropy and free enthalpy at different temperatures The heat of hydration of copper sulfate Thermal analysis Acidity of solutions. Conductivity measurement Reaction rate. The kinetic of simple and complex reactions Activity series of metals	Teaching methods Using and organising techniques, apparatus and materials; Observing, measuring and recording; Handling experimental observations and data; Planning and evaluating	Notes Mathematic al modeling and numerical simulations, experimenta l apparatus.
8.2. 1 1 2 3 4 5 6 7 8 9 10 11 12 13 13	Applications (Laboratory)Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils, glassware and laboratory equipmentAcid-base titration. Determination by titration of acetic acid content of vinegar Determination of molar mass of carbon dioxideHydrated Ionic CompoundCaffeine isolationHydrolyzeDetermination of enthalpy, entropy and free enthalpy at different temperatures The heat of hydration of copper sulfateThermal analysisAcidity of solutions. Conductivity measurement Reaction rate. The kinetic of simple and complex reactions Activity series of metals Cu spontaneous deposition. Protection of metals against corrosion	Teaching methods Using and organising techniques, apparatus and materials; Observing, measuring and recording; Handling experimental observations and data; Planning and evaluating investigations.	Notes Mathematic al modeling and numerical simulations, experimenta l apparatus.
8.2. 4 1 2 3 4 5 6 7 8 9 10 11 12 13 14	Applications (Laboratory) Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils, glassware and laboratory equipment Acid-base titration. Determination by titration of acetic acid content of vinegar Determination of molar mass of carbon dioxide Hydrated Ionic Compound Caffeine isolation Hydrolyze Determination of enthalpy, entropy and free enthalpy at different temperatures The heat of hydration of copper sulfate Thermal analysis Acidity of solutions. Conductivity measurement Reaction rate. The kinetic of simple and complex reactions Activity series of metals Cu spontaneous deposition. Protection of metals against corrosion Metal corrosion	Teaching methods Using and organising techniques, apparatus and materials; Observing, measuring and recording; Handling experimental observations and data; Planning and evaluating investigations.	Notes Mathematic al modeling and numerical simulations, experimenta l apparatus.
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8.2. <i>A</i> 1 2 3 4 5 6 7 8 9 10 11 12 13 14 Biblid 1. M Cluj-i 2. A. 5, ISI 3. L. 2009,	 Applications (Laboratory) Presentation of work. Safety norms. Analytical balance. Chemical laboratory utensils, glassware and laboratory equipment Acid-base titration. Determination by titration of acetic acid content of vinegar Determination of molar mass of carbon dioxide Hydrated Ionic Compound Caffeine isolation Hydrolyze Determination of enthalpy, entropy and free enthalpy at different temperatures The heat of hydration of copper sulfate Thermal analysis Acidity of solutions. Conductivity measurement Reaction rate. The kinetic of simple and complex reactions Activity series of metals Cu spontaneous deposition. Protection of metals against corrosion Metal corrosion ography -L. Ungureşan, L. Jantschi, D. Gligor, <i>Educational Applications of Chemistry</i> Napoca, 2004. Mesaroş, L. Bolunduţ, ML. Ungureşan, <i>General Chemistry Experiments</i>, Ed. BN: 978-973-141-228-3, 2010, pg. 197. Bolunduţ, A. Mesaroş, ML. Ungureşan, <i>Electrochemistry Experiments</i>, Ed. Ga, pg. 110. 	Teaching methods Using and organising techniques, apparatus and materials; Observing, measuring and recording; Handling experimental observations and data; Planning and evaluating investigations.	Notes Mathematic al modeling and numerical simulations, experimenta l apparatus. d. Mediamira, Colecția Tehne ecția Tehne 1,
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9. Bridging course contents with the expectations of the representatives of the community, professional associations and employers in the field
 Collaborations with: INCDTIM Cluj, Faculty of Chemistry and Chemical Engineering, UBB Cluj, Faculty of Environmental Science and Engineering UBB.

10. Evaluation

Activity type	10.1	Assessment criteria	10.2	Assessment methods	10.3	Weight in the final grade
Course		Written Examination		Multiple choice		80%
				evaluation - 2 nr.		

Applications		Laboratory test		The written test -1h		20%	
10.4 Minimum standard of performance							
• Exam grade ≥ 5							
• Laboratory grade > 5							

Course responsible Assoc. prof. chem. Mihaela Unguresan Head of department Prof.dr.eng. Rodica Potolea